Hurlstone Ag. H.S. Chemistry HSC Assessment Task #3 2005

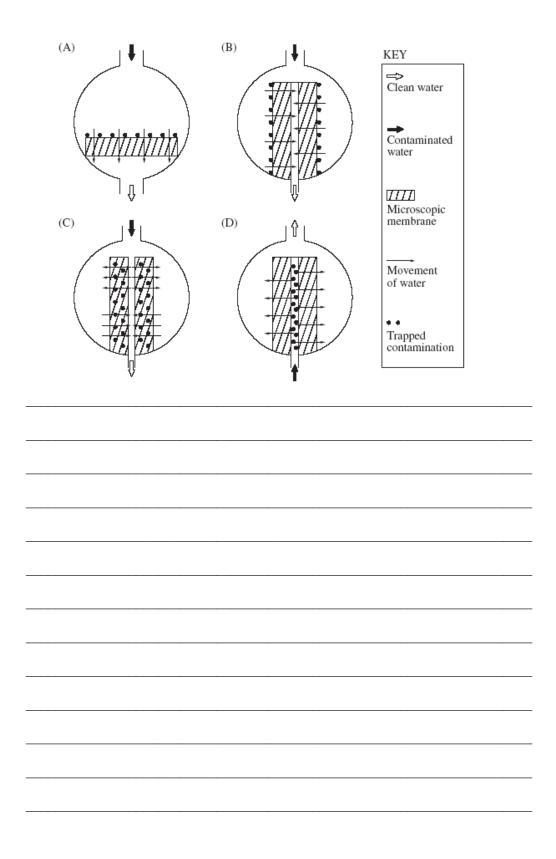
Student Number:		
Time allowed: 45 minutes Total Marks: 30 marks Answer the questions in the space provided. The value of each question is question.	given in brackets at the end of e	ach
1. a) Discuss the problems associated with the use of CFC's.	(5 marks)	
h) Outling the stone that have been taken to allowing these much laws	(2 montrs)	
b) Outline the steps that have been taken to alleviate these problems.	(2 marks)	

(1 mark)

2. a)Draw the structural formula of 1,2-dichloro-1,1,2,2-tetrafluoroethane.

Draw the structural formula of an isomer of 1,2-dichloro-1,1,2,2-tetrafluoroethane and na (2 marks)	
c) Write an equation to show what would happen when 1,2-dichloro-1,1,2,2-tetrafluoroethane is exposed to UV light.	
3. Describe how information regarding atmospheric concentrations of ozone are obtained. (2 marks)	
4. a) Identify a chemical reaction that needs monitoring and justify the need to monitor it. (5 marks)	

	
	
	
	
	
5. You were asked to present information from secondary sources to identify replace CFCs and evaluate the their use as a replacement for CFCs. Description the data you collected regarding this point. (2 marks)	fy alternative chemical used to be how you assessed the reliability
	
6. The diagram below shows 4 different designs of microscopic membrane	
of effectiveness and justify your rankings. (5 marks))



7. The following diagram summarises an example of the processes involved in treating water to make it suitable for drinking. There are 4 major processes illustrated (labelled 1-4). Identify each process and outline its purpose.

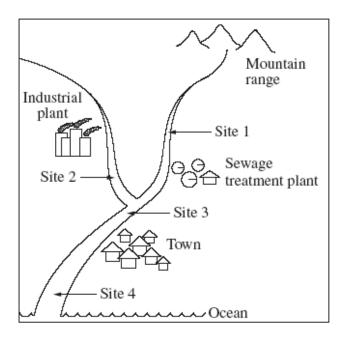
(6 marks)



Addition of iron chloride

 Cl_2

The map shows a catchment area. There is an industrial plant, a sewage treatment plant and a small town, all of which discharge water into the river. Water samples were collected at four sites.



The results of analysis of cadmium levels from these four sites are given in the table.

Sample site	Absorbance
Site 1	0.08
Site 2	0.15
Site 3	0.55
Site 4	0.40

(b) Justify your conclusion about the most likely source of cadmium pollution.

Chemistry Assessment Task 3 Research task – 2005- Marking Criteria

1. (a)

Criteria	Marks
Identifies the major issue as 'CFC's destroying ozone in the upper atmosphere'	4 - 5 marks
linking this to the subsequent increase in levels of harmful UV radiation reaching the Earth's surface	
AND	
Describes TWO or THREE disadvantages of these increased levels of UV radiation.	
Identifies the major issue as 'CFC's destroying ozone in the upper atmosphere'	1 - 3 marks
linking this to the subsequent increase in levels of harmful UV radiation reaching the Earth's surface	
AND/OR	
Describes ONE, TWO or THREE disadvantages of increased levels of UV radiation reaching the Earth's surface	

1.(b)

Criteria	Marks
Outlines the international agreement to phase out the use of CFC's (the Montreal protocol)	2 marks
AND EITHER	
Outlines at least one alternative available for use instead of CFC's	
OR	
Describes other ways to prevent the effects of exposure to harmful UV radiation	
One of the three points listed above	

2. (a)

Criteria	Marks
Draws the structural formula for 1,1,2,2-tetrachloro-1,2-difluoroethane.	1 mark

2.(b)

Criteria	Marks
Draws the structural formula for 1,1,1,2-tetrachloro-2,2-difluoroethane	1 mark
AND	
Names it.	

2.(c)

Criteria			Marks
Writes a correct equation, s unpaired electron).	howing the forn	nation of the chlorine free radical (an atom that has an	3 marks
	UV radiation		
CCI ₂ FCCI ₂ F	\rightarrow	CCl₂FCCIF + CI	

3.

Criteria	Marks
Identify that a spectrophotometer is the device used to collect information (which may be land-based, attached to a helium balloon or located on a satellite)	2 marks
AND	
Describes how the spectrophotometer collects relevant information (reference to wavelength intensities, absorption by ozone, etc)	
One of the two points above	1 mark

4.

Criteria	Marks
Identifies a chemical reaction that needs to be monitored, using appropriate and relevant equations to outline the chemical reaction(s)	4 - 5 marks
AND	
thoroughly describes TWO or THREE specific reasons why it is of importance to monitor the reaction as justification	
Identifies a chemical reaction that needs to monitored	2 – 3 marks
AND	
Uses appropriate and relevant equations to outline the chemical reaction(s)	
AND/OR	
Justifies the need for the reaction to be monitored by identifying ONE, TWO or THREE specific reasons why it is of importance to monitor the reaction (eg. production of pollutants, reduced energy output, reduced yield of desired product)	
Identifies a reaction that requires monitoring	1 mark

5.

Criteria	Marks
Identifies that reliability of information is ensured by collecting and analysing information from many sources (especially from highly reputable, unbiased sources such as government research bodies, eg. CSIRO) ensuring that only consistent information is used (this makes it reliable)	2 marks
Identifies that information should be collected from a number of sources to make it more reliable.	1 mark

6.

Criteria	Marks
Correct rank and appropriate justification for the relative position of all designs	5 marks
Correct rank and appropriate justification for the relative position of 3 designs	4 marks
Correct rank and appropriate justification for the relative position of 2 designs or appropriate justification for the relative position of 3 designs	3 marks
Correct rank and appropriate justification for the relative position of 1 design or appropriate justification for the relative position of 2 designs	2 marks
Correct rank or appropriate justification for the relative position of 1 designs	1 mark

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Criteria	Marks
Correctly identifies each process and outlines the role of each.	6 marks
½ marks were deducted for each process not correctly identified and 1 mark was deducted for each process whose purpose was not correctly outlined.	1- 5.5 marks