## Sydney Grammar School



# 2014 <br> FORM VI <br> TRIAL HSC EXAMINATION 

## Physics

Thursday $7^{\text {th }}$ August 8.40 a.m.

## General Instructions

- Reading time - 5 minutes
- Working time -3 hours
- Write using blue or black pen
- Draw diagrams using pencil
- Board-approved calculators may be used
- A data sheet, formulae sheets and periodic table are provided at the back of this paper
- Write your candidate number at the top of each page in Part B
- Hand in the paper in ONE bundle at the end of the exam.


## Check List

Each candidate must have

- Question paper
- Multiple choice answer sheet
- $2 \times$ Five-page booklets

Total marks (100)
Section I Pages 3-28
(75 marks)
This section has two parts, Part A and Part B

Part A - 20 marks

- Attempt Questions 1-20
- Allow about 35 minutes for this part

Part B-55 marks

- Attempt All Questions
- Allow about 1 hour and 40 minutes for this part

Section II Pages 29-31
(25 marks)

- Use separate writing booklets
- Attempt Question 38 only.
- Allow about 45 minutes for this section


## Masters

AAH - Dr A. Haines
SRW - Mr S. Williams
MRW - Dr M. Ward

1 If the mass of the Earth was doubled and its radius was also doubled then the new value for the acceleration due to gravity on Earth would be:
(A) 0.5 g
(B) g
(C) 2 g
(D) 4 g

2 If $\mathrm{r}_{1}$ and $\mathrm{T}_{1}$ represent the distance and period of Planet 1 from a star of mass M and $r_{2}$ and $T_{2}$ represent the distance and period of Planet 2 from the same star, then the quantity $\left(\frac{r_{1}^{3}}{T_{1}^{2}}\right) \times\left(\frac{T_{2}^{2}}{r_{2}^{3}}\right)$ would equal:
(A) M
(B) $\mathrm{GM} /\left(4 \pi^{2}\right)$
(C) 1
(D) $4 \pi^{2} /(\mathrm{GM})$

3 A projectile fired at a speed of $u$ and an angle of $\theta$ to the horizontal, reaches a maximum height of $H$. When the same projectile is fired again at the same angle with a new higher initial speed it reaches a new maximum height of 2 H . It true to say that:
(A) the new initial speed was $2 u$
(B) the new initial speed was $\sqrt{2} u$
(C) the new initial speed was $4 u$
(D) the new initial speed was $u / 2$

4 A train travelling at 0.8 c passes an observer on the platform who measures the length of the train as 100 m . What would a passenger on the train measure the length of the train to be?
(A) 60 m
(B) 100 m
(C) 125 m
(D) 167 m

5 An experiment was performed to investigate how the Period T of a pendulum varies with its length $L$. Theoretically the square of the period of a pendulum is given by the formula $\mathrm{T}^{2}=4 \pi^{2} \frac{L}{g}$. If the experimental results were graphed with the independent variable on the horizontal axis, which graph best represents the relationship between the variables?
(A)
L

(B)

(C)
L

(D)


6 A satellite of mass $m$ in orbit around the Earth has a velocity of $v$. The gravitational potential energy of the satellite is:
(A) $-m v^{2}$
(B) $-\frac{G m}{r v}$
(C) $-\frac{G m v}{r}$
(D) $-\frac{1}{2} m v^{2}$

7 A bomber is flying horizontally at $200 \mathrm{~ms}^{-1}$ at an altitude of 120 m . At what horizontal distance from the target does the bomber need to drop a bomb so that it hits the target?

(A) 700 m
(B) 767 m
(C) 990 m
(D) 4898 m

8 An electric motor with a constant voltage supply is used to raise a weight. If the weight is decreased the speed of rotation of the motor increases. Which of the following quantities will decrease in value?
(A) Resistance of the coil
(B) Back emf induced in the coil
(C) Current in coil
(D) The rate of change of flux in the coil

9 In the graph shown, the solid curve shows how the emf produced by a simple generator varies with time. The dashed curve is the output from the same generator after a modification has been made to the generator.


Which modification was made to produce the result shown?
(A) The speed of rotation of the coil was doubled.
(B) The number of turns in the coil was doubled.
(C) A split-ring commutator was added.
(D) The area of the coil was halved.

10 Two parallel wires have identical currents running through them. If the wires are 0.15 m apart and attract each other with a force of $2.5 \times 10^{-6} \mathrm{~N}$ per metre, what is the current in each wire?
(A) $\quad 0.53 \mathrm{~A}$
(B) $\quad 1.37 \mathrm{~A}$
(C) $\quad 1.88 \mathrm{~A}$
(D) $\quad 9.13 \mathrm{~A}$

11 The diagram below shows a side-on view of a square coil in a simplified DC motor. The coil has sides of length 0.250 m and consists of 100 turns. The magnitude of the uniform magnetic field strength between the poles is 0.800 T .


At the instant shown, if the magnitude of the torque is 0.775 Nm , what is the current in the coil?
(A) $\quad 0.0603 \mathrm{~A}$
(B) $\quad 0.101 \mathrm{~A}$
(C) $\quad 0.202 \mathrm{~A}$
(D) $\quad 0.241 \mathrm{~A}$

12 A diagram of a loudspeaker is shown below.


Which row in the following table shows the set of conditions that would result in the loudspeaker producing a loud high pitched sound?

|  | AC voltage frequency | AC voltage |
| :--- | :---: | :---: |
| (A) | Low | Low |
| (B) | Low | High |
| (C) | High | Low |
| (D) | High | High |

13 The primary coil of an ideal transformer is connected to a voltage supply $\left(\mathrm{V}_{\mathrm{p}}\right)$. This input voltage is varied and the corresponding voltage output in the secondary coil $\left(\mathrm{V}_{\mathrm{s}}\right)$ is recorded. The following graph shows the results.


The number of turns in the secondary coil is 120 .
What is the number of turns in the primary coil?
(A) 10
(B) 100
(C) 144
(D) 14400

14 Two parallel horizontal plates are separated by a distance of $d$ metres and have a potential difference of V volts maintained between them. Also present is a uniform magnetic field of B Teslas. A horizontal beam of electrons is directed between the plates so that it is moving at right angles to the magnetic field as shown below.


The electron beam passes between the plates undeflected.
Which choice below best denotes the experimental setup for this to occur?

|  | The speed of the electrons | Polarity of plate X with respect to plate Y |
| :---: | :---: | :---: |
| (A) | $\mathrm{v}=\mathrm{E} / \mathrm{B}$ | positive |
| (B) | $\mathrm{v}=\mathrm{B} / \mathrm{E}$ | positive |
| (C) | $\mathrm{v}=\mathrm{E} / \mathrm{B}$ | negative |
| (D) | $\mathrm{v}=\mathrm{B} / \mathrm{E}$ | negative |

15 The diagram below shows a charged particle positioned between two charged, parallel, metal plates.


Using the information given in the diagram, determine the magnitude of the electrostatic force acting on the particle.
(A) $1.4 \times 10^{-5} \mathrm{~N}$
(B) $7.5 \times 10^{-1} \mathrm{~N}$
(C) $2.8 \times 10^{5} \mathrm{~N}$
(D) $\quad 1.9 \times 10^{6} \mathrm{~N}$

16 In the field of Quantum Physics, Max Planck is known for:
(A) mass defect and the binding energy of a nucleus.
(B) the concept of wave-particle duality.
(C) the explanation of the black-body spectrum.
(D) the first quantum mechanical model of the Hydrogen atom.

17 A certain type of infra-red laser emits radiation with a wavelength of 930 nm . What frequency does this correspond to?
(A) $2.8 \times 10^{11} \mathrm{~Hz}$
(B) $3.2 \times 10^{14} \mathrm{~Hz}$
(C) $5.8 \times 10^{21} \mathrm{~Hz}$
(D) $1.4 \times 10^{27} \mathrm{~Hz}$

18 Which of the diagrams below best represents the electric field around two opposite charges?
(A)

(B)

(C)

(D)


19 Consider the diagram below, which shows a striation pattern in a cathode ray tube.


What is the cause of striation patterns like this?
(A) High voltages across low pressure gases.
(B) The interference of radio waves.
(C) Black-body radiation.
(D) The photoelectric effect.

20 The diagram below shows an electron moving in a uniform magnetic field.


If the magnitude of the force acting on the electron is $1.4 \times 10^{-14} \mathrm{~N}$, calculate its speed.
(A) $1.2 \times 10^{4} \mathrm{~ms}^{-1}$
(B) $2.3 \times 10^{4} \mathrm{~ms}^{-1}$
(C) $2.7 \times 10^{4} \mathrm{~ms}^{-1}$
(D) $4.6 \times 10^{4} \mathrm{~ms}^{-1}$

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## Candidate Number:

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## Part B

Total marks - 55
Attempt Questions 21-34
Allow about 1 hour and 40 minutes for this Part
Answer the questions in the spaces provided.
Show all relevant working in questions involving calculations.

Question 21 (4 marks)

A dart player releases a dart 2.4 m away from the dartboard where the bullseye is 1.7 m from the ground. The player successfully hits the bullseye by throwing from a height of 1.5 m at an angle of $30^{\circ}$ above the horizontal.


Determine the intial velocity of the dart
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## Question 22 (2 marks)

Marks

The graph below shows the minimum rocket energy needed per kilogram of mass to reach an orbital altitude of 250 km as a function of launch latitude on the Earth's surface.


With reference to this graph, discuss the effect of the Earth's motion on the launch of a rocket.
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## Candidate Number:

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Question 23 (4 marks)
Marks

A projectile of mass 1150 kg is fired vertically from the surface of an asteroid of mass $1.1 \times 10^{20} \mathrm{~kg}$ and radius $2.6 \times 10^{6} \mathrm{~m}$ with a speed of $50 \mathrm{~ms}^{-1}$.
a) Determine the initial kinetic energy of the projectile.
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$\qquad$
b) Determine the initial gravitational potential energy of the projectile.
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c) Determine the maximum distance the projectile reaches from the centre of the asteroid.
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## Candidate Number:

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Question 24 (5 marks)
Marks

A satellite of mass 1200 kg is in orbit around the Earth at a distance of 22000 km from the centre of the Earth.
a) Calculate the magnitude of the centripetal acceleration of the satellite at this distance.
$\qquad$
b) Determine the period of the satellite.
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c) Explain why the orbital velocity is independent of the mass of the satellite.
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## Candidate Number:

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Question 25 (3 marks)

A laser mounted on the floor of a train carriage travelling at high speed fires a light pulse towards a mirror on the ceiling of the train directly above the laser. An observer in the carriage can time how long this light pulse takes to travel from the laser, reflect from the mirror on the ceiling and then travel back to a detector at the laser source.
With the aid of a diagram and with reference to an external observer, explain how this situation demonstrates the phenomenon of time dilation.
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Candidate Number:

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## Candidate Number:

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Question 26 (3 marks)
Marks

A straight wire with a current of 5.0 A is placed in a uniform magnetic field of 0.10 T as shown in the diagram below.


Calculate the magnitude and direction of the magnetic force on a 5.0 cm length of the straight wire.
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Question 27 (2 marks)

Name one design strategy that is used to reduce heating in transformers and explain why this strategy is effective.
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## Candidate Number:

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Question 28 (4 marks)
Marks

A rectangular single turn loop of wire PQRS , with a resistance of $0.6 \Omega$, is placed at right angles to a uniform magnetic field of 0.15 T directed into the page. The width of the loop PS is 10 cm and the length PQ is 20 cm as shown in the following diagram.

$\mathrm{B}=0.15 \mathrm{~T}$ into the page

The magnetic field is decreased uniformly to zero in 0.03 s .
a) Calculate the magnitude of the current induced in the loop.
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b) On the diagram above, clearly indicate with an arrow the direction of the current induced in the loop.

## Candidate Number:

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Question 29 (5 marks)
Marks

A permanent magnet is moved very quickly in an anti-clockwise direction above an aluminium pie dish that is floating on water as shown in the following diagram.

a) Explain using physical laws what happens to the pie dish when the magnet is moving above the pie dish.
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b) This experiment is used to demonstrate the principal of an AC induction motor. Identify the part of an AC induction motor represented by the pie dish.
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Question 30 (5 marks) Marks

The diagram below shows a simple generator used in a Physics practical lesson.

a) A student notices that when the switch is closed and the light globe glows, the handle becomes harder to turn than when the switch is open and the light globe is not glowing. Explain the students observations.
$\qquad$
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b) This generator has a split ring/commutator to connect the rotating coil to the light globe. On the following axes, draw a graph of how the current in the light globe would change over time when the handle is rotated at a steady rate for two complete turns. Assume the coil of the generator is at the position shown in the diagram above at $\mathrm{t}=0 \mathrm{~s}$.


Question 31 (4 marks)

With the aid of a diagram describe how Heinrich Hertz demonstrated that radio waves were part of the electromagnetic spectrum.
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## Candidate Number:

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Question 32 (4 marks)

A cathode ray oscilloscope is a modified version of the cathode ray tube. Justify this statement, with reference to the function of a cathode ray oscilloscope and the features of a cathode ray tube.
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## Candidate Number:

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Question 33 ( 6 marks)

Assess the role of experiments that were conducted to determine the nature of cathode rays in the late nineteenth century.
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## Candidate Number:

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Question 34 (4 marks) Marks

Sodium metal has a threshold frequency of $5.50 \times 10^{14} \mathrm{~Hz}$. When low intensity blue light of frequency $7.60 \times 10^{14} \mathrm{~Hz}$ is shone onto the sodium, electrons are emitted from it. However, when high intensity red light of frequency $4.30 \times 10^{14} \mathrm{~Hz}$ is shone onto the metal, no electrons are emitted.
a) Calculate the work function of sodium.
$\qquad$
$\qquad$
b) Calculate the maximum kinetic energy of the electrons emitted from the sodium by the blue light above.
$\qquad$
$\qquad$
c) With reference to Einstein's photon model of light, explain why high intensity red light will not emit electrons from sodium, but low intensity blue light will.
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## Section II

25 marks
Attempt Question 38 only from Questions 35-39
Allow about 45 minutes for this section.
Answer the question in separate writing booklets. Extra writing booklets are available.
Show all relevant working in questions involving calculations.

## Pages

Question $35 \quad$ Elective 1
Question 36 Elective 2
Question 37 Elective 3
Question 38 From Quanta to Quarks................... 30-31
Question 39 Elective 5

## Candidate Number:

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Question 38 - From Quanta to Quarks ( 25 marks)
a) Outline the features of the Rutherford model of the atom.
b) For the Hydrogen atom:
(i) calculate the longest wavelength in the Balmer series of spectral emission lines.
(ii) determine the energy difference between the stationary states that produced this emission line.
(iii) describe how Bohr's model of the hydrogen atom was able to explain the emission spectrum of hydrogen.
c) In the treatment of cancers by radiotherapy a metastable Barium-137 isotope ( $\mathrm{Ba}-137 \mathrm{~m}$ ) is used to produce a gamma ray of energy 662 keV . Determine the change in binding energy per nucleon for this process.

## Question 38 continued on next page

## Candidate Number:

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## Question 38 - From Quanta to Quarks (continued)

## START A NEW BOOKLET FOR PARTS d), e) AND f)

d) The isotope Dysprosium-150 decays $64 \%$ of the time by $\beta^{+}$decay, and $36 \%$ of the time by $\alpha$ decay.
(i) The equation for the $\alpha$ decay is

$$
{ }_{66}^{150} D y \rightarrow{ }_{64}^{146} G d+{ }_{2}^{4} \alpha
$$

Calculate the energy released in this $\alpha$ decay using the following data table:

| Isotope | Nuclear Mass <br> (atomic mass units) |
| :---: | :---: |
| ${ }_{66}^{150} \mathrm{Dy}$ | 149.925585 |
| ${ }_{64}^{146} G d$ | 145.918311 |
| ${ }_{2}^{4} \alpha$ | 4.001506 |

(ii) Write the transmutation equation for the $\beta^{+}$decay.
e) Explain how the Davisson and Germer experiment supported Louis de Broglie's idea of matter waves.
f) Describe how the principles of conservation of momentum and conservation of energy were used in the discovery of the neutrino by Pauli, and the neutron by Chadwick.

## End of Paper

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## Physics

## Data Sheet

Charge on the electron, $q_{e}$
Mass of electron, $m_{e}$
Mass of neutron, $m_{n}$
Mass of proton, $m_{p}$
Speed of sound in air
Earth's gravitational acceleration, $g$
Radius of Earth, $R_{E}$
Mass of Earth
Speed of light, $c$
Magnetic force constant, $\left(k \equiv \frac{\mu_{0}}{2 \pi}\right)$
Universal gravitational constant, $G$
Planck's constant, $h$
Rydberg's constant, $R_{\mathrm{H}}$
Atomic mass unit, $u$

1 eV
Density of water, $\rho$
Specific heat capacity of water
$-1.602 \times 10^{-19} \mathrm{C}$
$9.109 \times 10^{-31} \mathrm{~kg}$
$1.675 \times 10^{-27} \mathrm{~kg}$
$1.673 \times 10^{-27} \mathrm{~kg}$
$340 \mathrm{~m} \mathrm{~s}^{-1}$
$9.8 \mathrm{~m} \mathrm{~s}^{-2}$
$6.4 \times 10^{6} \mathrm{~m}$
$6.0 \times 10^{24} \mathrm{~kg}$
$3.00 \times 10^{8} \mathrm{~m} \mathrm{~s}^{-1}$
$2.0 \times 10^{-7} \mathrm{~N} \mathrm{~A}^{-2}$
$6.67 \times 10^{-11} \mathrm{~N} \mathrm{~m}^{2} \mathrm{~kg}^{-2}$
$6.626 \times 10^{-34} \mathrm{~J} \mathrm{~s}$
$1.097 \times 10^{7} \mathrm{~m}^{-1}$
$1.661 \times 10^{-27} \mathrm{~kg}$
$931.5 \mathrm{MeV} / \mathrm{c}^{2}$
$1.602 \times 10^{-19} \mathrm{~J}$
$1.00 \times 10^{3} \mathrm{~kg} \mathrm{~m}^{-3}$
$4.18 \times 10^{3} \mathrm{~J} \mathrm{~kg}^{-1} \mathrm{~K}^{-1}$

## FORMULAE SHEET

$$
\begin{aligned}
& v=f \lambda \\
& I \propto \frac{1}{d^{2}} \\
& E_{p}=-\frac{G m_{1} m_{2}}{r} \\
& F=m g \\
& \frac{v_{1}}{v_{2}}=\frac{\sin i}{\sin r} \\
& E=\frac{F}{q} \\
& R=\frac{V}{I} \\
& P=V I \\
& v_{y}^{2}=u_{y}^{2}+2 a_{y} \Delta y \\
& \text { Energy }=\text { VIt } \\
& v_{a v}=\frac{\Delta r}{\Delta t} \\
& a_{a v}=\frac{\Delta v}{\Delta t}=\frac{v-u}{t} \\
& F=\frac{G m_{1} m_{2}}{d^{2}} \\
& \sum F=m a \\
& E=m c^{2} \\
& F=\frac{m \nu^{2}}{r} \\
& l_{v}=l_{0} \sqrt{1-\frac{v^{2}}{c^{2}}} \\
& E_{k}=\frac{1}{2} m v^{2} \\
& W=F s \\
& p=m v \\
& \text { Impulse }=F t \\
& m_{v}=\frac{m_{0}}{\sqrt{1-\frac{v^{2}}{c^{2}}}}
\end{aligned}
$$

## FORMULAE SHEET

$$
\frac{F}{l}=k \frac{I_{1} I_{2}}{d} \quad d=\frac{1}{P}
$$

$$
F=B I l \sin \theta
$$

$$
M=m-5 \log \left(\frac{d}{10}\right)
$$

$$
\tau=F d
$$

$$
\tau=n B I A \cos \theta
$$

$$
\frac{I_{A}}{I_{B}}=100^{\frac{\left(m_{B}-m_{A}\right)}{5}}
$$

$$
\frac{V_{p}}{V_{s}}=\frac{n_{p}}{n_{s}} \quad m_{1}+m_{2}=\frac{4 \pi^{2} r^{3}}{G T^{2}}
$$

$$
F=q v B \sin \theta \quad \frac{1}{\lambda}=R_{H}\left\lfloor\frac{1}{n_{f}^{2}}-\frac{1}{n_{i}^{2}}\right\rfloor
$$

$$
E=\frac{V}{d}
$$

$$
\lambda=\frac{h}{m v}
$$

$$
E=h f
$$

$$
A_{0}=\frac{V_{\text {out }}}{V_{\text {in }}}
$$

$$
c=f \lambda
$$

$$
\frac{V_{\text {out }}}{V_{\text {in }}}=-\frac{R_{f}}{R_{i}}
$$

$$
Z=\rho v
$$

$$
\frac{I_{r}}{I_{0}}=\frac{\left[Z_{2}-Z_{1}\right]^{2}}{\left[Z_{2}+Z_{1}\right]^{2}}
$$

Sydney Grammar School


## 2014

HIGHER SCHOOL CERTIFICATE
TRIAL EXAMINATION
Thu 7 August 8.40 am

## General'Instructions

- Write your class and candidate number in the space provided.
- Attempt all questions 1 - 20
- Use a blue or black pen
- Select the alternative A, B, C, or D
that best answers the question.
- Fill in the response circle completely.



## Physics

 Section I Part A ANSWER SHEET
2. A
(B)
(D)
4. A
(B)
(C)
5. A
(B)
C)

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Candidate Number:

Part B
Total marks - 55
Attempt Questions 21-34
Allow about 1 hour and 40 minutes for this Part
Answer the questions in the spaces provided.
Show all relevant working in questions involving calculations.


Question 21 (4 marks)

A dart player releases a dart 2.4 m away from the dartboard where the bullseye is 1.7 m from the ground. The player successfully hits the bullseye by throwing from a height of 1.5 m at an angle of $30^{\circ}$ above the horizontal.


Determine the intial velocity of the dart

$\qquad$
$\qquad$


## Candidate Number:

$\qquad$
Question 22 (2 marks)

The graph below shows the minimum rocket energy needed per kilogram of mass to reach an orbital altitude of 250 km as a function of launch latitude on the Earth's surface.


With reference to this graph, discuss the effect of the Earth's motion on the launch of a rocket.



## Candidate Number:

$\qquad$

Question 23 (4 marks)

A projectile of mass 1150 kg is fired vertically from the surface of an asteroid of mass $1.1 \times 10^{20} \mathrm{~kg}$ and radius $2.6 \times 10^{6} \mathrm{~m}$ with a speed of $50 \mathrm{~ms}^{-1}$.
a) Determine the initial kinetic energy of the projectile.

b) Determine the initial gravitational potential energy of the projectile.

$$
-3.245 \times 10^{6}-\mathrm{J}
$$

c) Determine the maximum distance the projectile reaches from the centre of the asteroid.



Candidate Number: $\qquad$

Question 24 (5 marks)

A satellite of mass 1200 kg is in orbit around the Earth at a distance of 22000 km from the centre of the Earth.
a) Calculate the magnitude of the centripetal acceleration of the satellite at this distance.

b) Determine the period of the satellite.

c) Explain why the orbital velocity is independent of the mass of the


Candidate Number: $\qquad$

Question 25 (3 marks)
Marks

A laser mounted on the floor of a train carriage travelling at high speed fires a light pulse towards a mirror on the ceiling of the train directly above the laser. An observer in the carriage can time how long this light pulse takes to travel from the laser, reflect from the mirror on the ceiling and then travel back to a detector at the laser source.
With the aid of a diagram and with reference to an external observer, explain how this situation demonstrates the phenomenon of time dilation.


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## Candidate Number:

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Candidate Number: $\qquad$

Question 26 (3 marks)

A straight wire with a current of 5.0 A is placed in a uniform magnetic field of 0.10 T as shown in the diagram below.


Calculate the magnitude and direction of the magnetic force on a 5.0 cm length of the straight wire.

$$
\begin{aligned}
& F=n B I L \sin \theta \\
& =1 \times 0.10 \times 5.0 \times 5 / 100 \times \sin 90 \\
& =0.025 \mathrm{~N} \\
& \text { Notes: 1. If } F=2.5 \mathrm{~N}+\text { correct } 1 / 1 \times 2 \text { mks } \text { direction }^{2} \text { down the with page an }
\end{aligned}
$$

3. If correct substitution int Direction
correct formula but answer
incorrect with correct direction $\sqrt{ } / \times$ Zmbs

Question 27 (2 marks)

Name one design strategy that is used to reduce heating in transformers and explain why this strategy is effective.
Cork Name of strategy egg heat sinks, Laminationis
$V$ (1 )mk + correct explanation based on minimises eddy currents. . Minimises heat production $\because$ improves efficiency

## Candidate Number:

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Question 28 (4 marks)

A rectangular single turn loop of wire PQRS , with a resistance of $0.6 \Omega$, is placed at right angles to a uniform magnetic field of 0.15 T directed into the page. The width of the loop PS is 10 cm and the length PQ is 20 cm as shown in the following diagram.

$\mathrm{B}=0.15 \mathrm{~T}$ into the page

The magnetic field is decreased uniformly to zero in 0.03 s .
a) Calculate the magnitude of the current induced in the loop.
(1) mk for

$$
\begin{equation*}
\text { emf }=-1 \times(0.1 \times 0.2)[0.15-0] \tag{3}
\end{equation*}
$$

$\sqrt{ }$ (1)mk for
$I=e m f / R$
$\operatorname{emf}=-N \Delta \phi$ $\qquad$

$$
=0.167 \mathrm{~A}
$$

/Drink for
emf $=I R$. $\qquad$
b) On the diagram above, clearly indicate with an arrow the direction of the current induced in the loop.


Answer shown on
diagram.

Candidate Number: $\qquad$

Question 29 (5 marks)

A permanent magnet is moved very quickly in an anti-clockwise direction above an aluminium pie dish that is floating on water as shown in the following diagram.

To score
full marks,
answers
needed to
refe-to the physical laws 1.e. Faraday's Lam and

a) Explain using physical laws what happens to the pie dish when the magnet is moving above the pie dish.
(1) mk/ $M=$ pie dish rotates in the
same direction (ie anti-clockwise) as the magnet.
(1) Mk $\Delta \phi=$ moving magnet produces a
change in flux which by
(1) 1 FF $=$ Faraday's Law an emf is induced that results in eddy currents
(1) mo k

or $\Delta \phi \rightarrow I \rightarrow B \rightarrow$ interacts with the (motor effect)
force to oppose motion
b) This experiment is used to demonstrate the principal of an AC
induction motor. Identify the part of an AC induction motor represented by the pie dish.

died not allow
$\times$ coils
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PCK

## Candidate Number:

$\qquad$

Question 30 (5 marks)
Marks

The diagram below shows a simple generator used in a Physics practical lesson.

5. non-connected circuit is $\infty$ resistance not $0 \Omega$.

Notes:

1. emf's donot flow
$z$. extra effort is not due to overcoming
"back emf".
2. In Len 3 's Law,

- force opposes the motion "not the

4. do emit use the wren
"resistance" when you mean "force".
a) A student notices that when the switch is closed and the light globe glows, the handle becomes harder to turn than when the switch is open and the light globe is not glowing. Explain the students observations.
$\checkmark \begin{aligned} I & =\text { induced current flows when switch closed } \\ & =\text { no induced current flows when switch open }\end{aligned}$ $L=$ correct statement of Len's Lam ne direction of induced current produces a magnetic $F=$ force that opposes the motion of the coil OR/wher switch closed, current flows .more work needed to produce this energy: more force necessary as $u=$ Fr.
To obtain full marks, answers needed to refer to the physical principles involved and were logical and coherent and expressed in correct
b) This generator ias a split ring/commutator to connect the rotating coil to the light globe. On the following axes, draw a graph of how the current in the light globe would change over time when the handle is rotated at a steady rate for two complete turns. Assume the coil of the generator is at the position shown in the diagram above at $t=0 \mathrm{~s}$.


## AAH - CRIB

With the aid of a diagram describe how Heinrich Hertz demonstrated that radio waves were part of the electromagnetic spectrum.

## For FOUR MARKS:

- Provides a good, labelled diagram of the experimental set-up.
- Describes Hertz's production and reception of radio waves.
- Describes Hertz's investigation into the wave nature of the radio waves, including his measurement of their speed.
- Relates the speed of radio waves and their wave properties to the electromagnetic spectrum.

THREE MARKS were awarded for reasonable answers that either omitted the measurement of speed, or were lacking in sufficient detail.

TWO MARKS for answers that a described the production and reception of radio waves well, but contained very little else.

ONE MARK for answering some aspect of the question; e.g. how Hertz produced radio waves.

## NOTE:

1. Hertz's work on the photoelectric effect is not really relevant here.
2. Hertz did not use a radio to detect radio waves!
3. Too many diagrams were very poor, and added nothing to the answer.

A cathode ray oscilloscope is a modified version of the cathode ray tube.
Justify this statement, with reference to the function of a cathode ray
oscilloscope and the features of a cathode ray tube.

FOUR MARKS for GOOD answers addressing the following points:

- The CRO as a diagnostic tool, allowing analysis of voltage v time.
- The function of the three main features of the CRT, i.e.:
- The electron gun to produce an electron beam.
- A clear description of the role of the metal deflection plates: horizontal deflection to create a time axis, vertical deflection determined by an input voltage.
- A fluorescent screen, calibrated to allow quantitative measurement.

THREE MARKS were usually awarded for either:

- GOOD descriptions of the parts of the CRT, that did not address the function of the CRO, or
- REASONABLE answers that lacked some detail, usually in the specific role of the deflection plates.

TWO MARKS were usually awarded for demonstrating a REASONABLE understanding of the features of a CRT that were lacking in specifics.

ONE MARKS was usually awarded for demonstrating a BASIC understanding of the features of a CRT.

## NOTE:

1. The easiest way to get full marks on this question was to give a labelled diagram of the features of the CRO, with a small amount of explanatory text for every feature. Very few boys did this.
2. Some boys did not know the function of a CRO
3. Many boys confused cathode ray tubes with discharge tubes.

Assess the role of experiments that were conducted to determine the nature of cathode rays in the late nineteenth century.

FIVE or SIX MARKS for a GOOD ANSWER that addresses most of the following points:

- DESCRIBES early experiments with cathode ray tubes to investigate the properties of cathode rays, e.g.:
- Maltese Cross - to demonstrate CR travel in straight lines.
- Paddle Wheel - to demonstrate CR have momentum.
- Deflected by Magnetic Fields - CR are particles with negative charge.
- Not Deflected by Electric Fields - CR are a form of em radiation.
...and clearly shows the link between the experiment, the observation and the nature of cathode rays.
- OUTLINES the apparent contradictory properties of cathode rays.
- DESCRIBES the measurement of $\mathrm{e} / \mathrm{m}$ by Thomson and recognises that this experiment - credited as the discovery of the electron - is key to resolving the debate over the nature of cathode rays.
- PROVIDES an overall assessment, recognising the key role played by experiments in determining the nature of cathode rays.

THREE or FOUR MARKS for a REASONABLE ANSWER that:

- PROVIDES a reasonably comprehensive description of the experiments used to determine the nature of cathode rays.
- ASSESSES the importance of these experiments.

ONE or TWO MARKS for describing one or two experiments concerning the nature of cathode rays.

Sodium metal has a threshold frequency of $5.50 \times 10^{14} \mathrm{~Hz}$. When low intensity blue light of frequency $7.60 \times 10^{14} \mathrm{~Hz}$ is shone onto the sodium, electrons are emitted from it. However, when high intensity red light of frequency $4.30 \times 10^{14} \mathrm{~Hz}$ is shone onto the metal, no electrons are emitted.
a) Calculate the work function of sodium.

$$
\phi=h . f_{o}=6.626 \times 10^{-34} \times 5.50 \times 10^{14}=\underline{3.64 \times 10^{-19} \mathrm{~J}}
$$

b) Calculate the maximum kinetic energy of the electrons emitted from the sodium by the blue light above.

$$
K E_{M A X}=h\left(f-f_{0}\right)=6.626 \times 10^{-34} \times\left(7.60 \times 10^{14}-5.50 \times 10^{14}\right)
$$

$$
=1.39 \times 10^{-19} \mathrm{~J}
$$

c) With reference to Einstein's photon model of light, explain why high intensity red light will not emit electrons from sodium, but low intensity blue light will.

## For TWO MARKS:

- Makes reference to the photon model of light.
- Identifies the lack of effect of intensity.
- Relates the frequency / energy of red and blue photons to the threshold frequency / workfunction of sodium to explain the observations given.
(The best answers referred to E=hf, but that was not essential for full marks)

ONE MARK for addressing only one or two of these points.
(A large number of boys did not refer to the blue or red light in their answer!)

|  | Question 38 - From Quanta to Quarks (25 marks) |  | PCK |
| :---: | :---: | :---: | :---: |
| a) | Outline the features of the Rutherford model of the atom. |  | 2 |
|  | Marking guide | Marks |  |
|  | Addresses 2 of the following <br> - Structure - Small nucleus with orbiting electrons (mostly empty space) <br> - Mass - The central nucleus is dense and contains most of the mass of the atom <br> - Charge - The nucleus is positively charged with orbiting negative electrons (many answers fail to identify the electron as negatively charged) | 2 |  |
|  | Addresses only 1 feature correctly | 1 |  |

b) For the Hydrogen atom:
(i) calculate the longest wavelength in the Balmer series of spectral emission lines. $\mathbf{2}$

The longest wavelength (is smallest frequency) are is therefore created by the smallest energy change $\Delta \mathrm{E}=\mathrm{hf}$ (between adjacent stationary states in the balmer series)

Balmer series means $n_{f}=2$, smallest energy change when $n_{i}=3$. Using the Rydberg formula
$\frac{1}{\lambda}=R\left\lfloor\frac{1}{n_{f}{ }^{2}}-\frac{1}{n_{i}{ }^{2}}\right\rfloor \quad \therefore \frac{1}{\lambda}=1.097 \times 10^{7}\left[\frac{1}{2^{2}}-\frac{1}{3^{2}}\right]=1,523,611 \quad \therefore \lambda=6.5633546 \times 10^{-7} \mathrm{~m}$
wavelength $=\mathbf{6 5 6 n m}($ or $6.56 \mathrm{e}-7 \mathrm{~m})\left(\right.$ wrong answer 365 nm if using $\left.\mathrm{n}_{\mathrm{i}}=\infty\right)$

| Marking guide | Marks |
| :--- | :---: |
| Correctly identifying Balmer $\mathrm{n}_{\mathrm{f}}=2$, AND longest wavelength <br> from $\mathrm{n}_{\mathrm{i}}=3$ AND solves the Rydberg equation (ie correctly <br> inverting wavelength) | 1 |
| ONLY correctly completing TWO of the above | 1 |

Markers Note: (for those who picked $n_{i}=6$ ) The Balmer series is all lines where $n_{f}=2$ and $n_{i}>2$,
not just the lines in the visible part of the spectrum
(ii) determine the energy difference between the stationary states that produced this emission line.
The energy difference between stationary states is the same as the energy of the photon emitted

$$
\begin{gathered}
E=h f=\frac{h c}{\lambda}=\frac{\left(6.626 \times 10^{-34}\right)\left(3 \times 10^{8}\right)}{6.5633546 \times 10^{-7}}=3.0286342 \times 10^{-19} J \\
\\
\left.\mathbf{3 . 0 3 \times 1 0 ^ { - 1 9 } J}(3 \text { sigfig }) \text { also } 1.9 \mathrm{eV}\right)
\end{gathered}
$$

| Marking guide | Marks |
| :--- | :---: |
| Correct answer in Joules or eV | 1 |

(iii) describe how Bohr's model of the hydrogen atom was able to explain the emission spectrum of hydrogen.

| Marking guide | Marks |
| :---: | :---: |
| High mark answers are able to specifically (not generally) link how Bohr's model and postulates are able to explain how all the wavelengths of light in the hydrogen emission spectrum are produced <br> Answers included: <br> Descriptions of Bohr's modal, specifically the 3 postulates <br> AND <br> Critical parts connecting postulates to production of spectral emission lines <br> 1. Stationary energy levels lead to discrete spectral lines ( $1^{\text {st }}$ postulate). Not a spectrum of wavelengths <br> 2. Clearly explaining how the spectral lines are formed ( $2^{\text {nd }}$ postulate)(explicitly)The transition from a high energy stationary state to a lower energy state emits a photon with the same energy and the wavelength of the photon is determined by Einstein/Plank's $\mathrm{E}=\mathrm{hf}$. Better answers clearly distinguished the following: <br> - Clearly distinguishing that spectral emission lines are produced when transitioning from a higher orbit to a lower orbit <br> - One transition between 2 orbits corresponds to one spectral line in the emission spectrum <br> - The initial and final orbits are the numbers in Rydberg equation <br> - Describing how the different series in the Hydrogen spectrum are formed (including diagrammatically) <br> 3. Bohr was able to use $3^{\text {rd }}$ postulate to derive the Rydberg equation (quoted) which allows calculation of all wavelengths of the hydrogen emission spectrum | 4-5 |
| Simply stating Bohr's postulates without relating them to the production of hydrogen emission spectral line wavelengths OR <br> General statements about fixed orbits and how electrons changing orbits emits or absorbs photons of equal energy | 2-3 |
| Basic information about the movement of electrons producing light | 1 |

Markers Notes: Too many answers focussed on just the visible part of the hydrogen spectrum as opposed to the whole spectrum. The hydrogen spectrum is all spectral lines produced by hydrogen regardless of how 'visible' they are. The Lyman (UV) [1906-1914] and Paschen (IR) [1908] series were known by the time of Bohr's model [1913].
c) In the treatment of cancers by radiotherapy a metastable Barium-137 isotope ( $\mathrm{Ba}-137 \mathrm{~m}$ ) is used to produce a gamma ray of energy 662 keV . Determine the change in binding energy per nucleon for this process.
The change in binding energy in the nucleus is released as the energy released in the gamma ray. Similar to Bohr's model.

| Marking guide | Marks |
| :--- | :---: |
| Correct answer | 1 |
| $662,000 \mathrm{eV}$ per 137 nucleons $=4832 \mathrm{eV} /$ nucleon or |  |
| $4.83 \mathrm{keV} /$ nucleon (3sigfig) |  |
| (or $7.74 \times 10^{-16}$ Joules) |  |

d) The isotope Dysprosium-150 decays $64 \%$ of the time by $\beta^{+}$decay, and $36 \%$ of the time by $\alpha$ decay.
(i) The equation for the $\alpha$ decay is

$$
{ }_{66}^{150} D y \rightarrow{ }_{64}^{146} G d+{ }_{2}^{4} \alpha
$$

Calculate the energy released in this $\alpha$ decay using the following data table:
Mass defect $=149.925585-(145.918311+4.001506)=\mathbf{0 . 0 0 5 7 6 8} \mathbf{a m u}$
$0.005768 \times 931.5=5.372892 \mathrm{MeV}=\mathbf{5 . 4} \mathbf{~ M e V}$
or $(0.005768) \times\left(1.661 \times 10^{-27}\right) \times\left(9 \times 10^{16}\right)=8.6225832 \times 10^{-13}=\mathbf{8 . 6 \times 1 0} 0^{-13} \mathbf{J}$

| Marking guide | Marks |
| :--- | :---: |
| Correct mass defect (amu or kg) | 1 |
| Correct answer in (Joules or eV) | 1 |

(ii) Write the transmutation equation for the $\beta^{+}$decay.

Answer

$$
{ }_{66}^{150} D y \rightarrow{ }_{65}^{150} \mathrm{~Tb}+{ }_{+1}^{0} \beta+{ }_{0}^{0} v
$$

| Marking guide | Marks |
| :--- | :---: |
| Conservation of mass number 150 | 1 |
| Missing one thing <br> $\bullet \quad$ Tb symbol and atomic number 65 <br> $\bullet \quad$ Beta positive particle (positron) correct symbol <br> $\bullet \quad$ Neutrino <br> $\quad$$\quad$ (ignored mistake of identifying it <br> incorrectly as an antineutrino) <br> $\quad$ (ignored ambiguous drawing of greek nu <br> $(v)$ with gamma symbol $(\gamma)$ ) | 1 |

## e) Explain how the Davisson and Germer experiment supported Louis de Broglie's idea of matter waves.

| Marking guide | Marks |
| :--- | :---: |
| Davison Germer Experiment was to experimentally prove <br> Broglie's idea of matter waves <br> De Broglie proposed that matter has an associated <br> wavelength given by the matter's momentum $\lambda=\frac{h}{m v}$.(must <br> be included, with description, not just writing the equation) | 1 |
| Description of Davison and Germer experiment <br> Correct description and details of how the experiment was <br> performed with an electron beam (known velocity), Ni <br> crystal and Electron detector. A labelled diagram can be <br> sufficient. | 1 |
| Correct details about how the interference pattern was <br> observed or measured | 1 |
| Diffraction/interference pattern was produced by <br> - Changing electron speed at a fixed angle (what they <br> did) <br> fixed electron speed and changing incident angle <br> (also acceptable) | 1 |
| Summary/Quality <br> Diffraction and interference are a property of waves and the <br> wavelength measured correspond to the theoretical value <br> proposed by de Broglie's equation |  |

Markers note: Wave-particle duality was a concept that was accepted later - specifically after Heisenbergs uncertainty principle. De Broglie was still thinking that electrons were particles with an associated wavelength or pilot wave. used in the discovery of the neutrino by Pauli, and the neutron by Chadwick.
Answers were expected to clearly describe how a specific elastic collision where momentum and kinetic energy is conserved was used to determine the existence and properties of new particles.

| Marking guide | Marks |
| :---: | :---: |
| Notes |  |
| Discovery of the Neutron <br> - Description of Chadwick's experimental setup <br> - Explanation of elastic scattering of neutron and protons in paraffin (and/or recoil of hydrogen/nitrogen gases) allowing properties of neutron to be determined. <br> - Description of properties of neutron discovered. (ie Determines unknown radiation/particle is neutral and roughly 1.1 times the mass of a proton) | 1-3 |
| Discovery of the Neutrino <br> - Description of the Kinetic Energy spectrum of electrons during beta decay (Answers must specifically address the spectrum of the kinetic energy of the beta particles. Answers that simply state that the momentum/energy after the collision was less or different than before the decay is not enough) <br> - Existence of neutrino proposed by Pauli as additional particle in elastic scattering (sum of momentum of recoiling daughter nuclei and beta and neutrino is zero) so that momentum and kinetic energy is conserved. <br> - Description of neutrino properties needed: near massless, chargeless. <br> (technically Pauli proposed the necessity of the neutrino, but it was actually directly experimentally identified later) | 1-3 |

Markers Notes:

- Many are getting the axes incorrect on the beta particle kinetic energy after decay (this could be ignored if the written explanation expressed the correct meaning.


